

The transformation of macrophyte-derived organic matter to methane relates to plant water and nutrient contents

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Abstract

Macrophyte detritus is one of the main sources of organic carbon (OC) in inland waters, and it is potentially available for methane (CH₄) production in anoxic bottom waters and sediments. However, the transformation of macrophyte-derived OC into CH₄ has not been studied systematically, thus its extent and relationship with macrophyte characteristics remains uncertain. We performed decomposition experiments of macrophyte detritus from 10 different species at anoxic conditions, in presence and absence of a freshwater sediment, in order to relate the extent and rate of CH₄ production to the detritus water content, C/N and C/P ratios. A significant fraction of the macrophyte OC was transformed to CH₄ (mean = 7.9%; range = 0–15.0%) during the 59-d incubation, and the mean total C loss to CO₂ and CH₄ was 17.3% (range = 1.3–32.7%). The transformation efficiency of macrophyte OC to CH₄ was significantly and positively related to the macrophyte water content, and negatively to its C/N and C/P ratios. The presence of sediment increased the transformation efficiency to CH₄ from an average of 4.0% (without sediment) to 11.8%, possibly due to physicochemical conditions favorable for CH₄ production (low redox potential, buffered pH) or because sediment particles facilitate biofilm formation. The relationship between macrophyte characteristics and CH₄ production can be used by future studies to model CH₄ emission in systems colonized by macrophytes. Furthermore, this study highlights that the extent to which macrophyte detritus is mixed with sediment also affects CH₄ production.

Inland waters are important sources of methane (CH₄), a greenhouse gas with a global warming potential 28 times higher than that of carbon dioxide (CO₂) at a 100 yr scale (IPCC 2014). Reservoirs, lakes and rivers emit about 103 Tg(CH₄) yr⁻¹ (Bastviken et al. 2011), wetlands about 115–284 Tg(CH₄) yr⁻¹ (Mitsch et al. 2013; Saunio et al. 2016), and wetlands, rivers, and lakes may collectively account for ca. 40% of the global CH₄ emissions (IPCC 2014). CH₄ is mainly produced during the anoxic decomposition of organic carbon (OC) in sediments, and it is strongly controlled by temperature and the supply and biodegradability of organic matter (Segers 1998; Bastviken 2009). Because of high temperatures and primary productivity,

CH₄ emission from inland waters can be especially high in the tropics (Tranvik et al. 2009; Yvon-Durocher et al. 2014).

Aquatic macrophytes are plants that grow in or close to water, and that are visible to the naked eye, including macroalgae, bryophytes, pteridophytes, and nonwoody angiosperms (Sculthorpe 1967). Macrophytes contribute to a significant part of the primary production in wetlands and the littoral zones of lakes and rivers (Wetzel 1964; Jeppesen et al. 1997; Silva et al. 2013). In particular, in tropical systems, macrophytes have a very high productivity (Westlake 1963; Silva et al. 2009). For example, Junk and Howard-Williams (1984) measured a maximum biomass doubling time of 9.4 d for the fast-growing tropical species *Eichhornia crassipes* (Eicc) in an Amazonian floodplain lake, and Westlake (1963) estimated that this species could have a maximum annual production of 15 kg (fresh weight) m⁻² before a severe decrease due to self-shading effects. Macrophyte detritus may consequently be a potentially large and important source of OC to aquatic systems, available for CH₄ production in bottom anoxic waters and sediments. Despite the large literature on the difference in decomposition rate between macrophytes in oxic conditions (Webster and

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Benfield 1986; Xie et al. 2004; Longhi et al. 2008), the extent and the speed at which macrophyte detritus can be transformed to CH₄ at anoxic conditions is still poorly understood.

Macrophytes exhibit a wide range of lability to microbial degradation, often related to their nutrient stoichiometry (i.e., the C/N and C/P ratios) or content of structural compounds (e.g., polysaccharides and lignins) (Enriquez et al. 1993; Chimney and Pietro 2006; Longhi et al. 2008). The need for structural tissues differs between vascular plant species according to their position in the water column (Etnier and Villani 2007; Hamann and Puijalón 2013; De Wilde et al. 2014) and also is very low for macroalgae (Kankaala et al. 2003; Dai et al. 2005). The leaf water content (and inversely, the leaf dry matter content) is often used as an indicator of the abundance of structural tissues because it relates to the relative proportion of mesophyll vs. structural compounds (Garnier and Laurent 1994; Elger and Willby 2003; Kazakou et al. 2006). Because of these differences in structural compound contents, macroalgae are supposed to be the most labile to microbial decomposition, followed by submerged and floating vascular plants, while emergent plants are least labile (Webster and Benfield 1986; Hart 2004; Chimney and Pietro 2006). Labile OC is expected to be readily decomposed also at anoxic conditions, and to sustain high CH₄ production rates; conversely, the decomposition of chemically more complex structural compounds might be limited by low hydrolysis and fermentation rates (Kristensen et al. 1995; Bastviken et al. 2003; Grasset et al. 2018). However, there are few studies comparing the transformation efficiency of macrophyte OC to CH₄ (Kankaala et al. 2003; Vizza et al. 2017; Grasset et al. 2018), and due to the low number of species investigated (usually less than 4), no relationship with macrophyte characteristics has been demonstrated. Thus, there is at present no systematic understanding of how much CH₄ the decomposition of different types of macrophytes generates. We hypothesized that in anoxic conditions, macrophytes with high water content and low C/N and C/P ratios decompose more quickly, and transform more OC into CH₄, than macrophytes with low water content and high C/N and C/P ratios.

The quantity of detrital macrophyte OC deposited onto the sediment and the extent to which it is mixed to sediment can differ widely between and within systems, and may modify the physicochemical conditions, which in turn exert a strong control on methanogenesis (Segers 1998; Bastviken 2009). For example, a high deposition of detrital OC in soils and sediments may lead to a low pH due to an accumulation of end products such as fatty acids or phenols, and thus limit methanogenesis (Williams and Crawford 1984; Magnusson 1993; Emilson et al. 2018). However, the quantitative effect of a high amount of macrophyte detritus on top of the sediment on CH₄ production has never been assessed. We hypothesized that the extent and the rate of CH₄ production derived from macrophyte OC are higher when mixed with a freshwater sediment.

To test these two hypotheses, we incubated at anoxic conditions for ca. 60 d senescent aboveground tissues from

10 macrophyte species of different life forms, in presence and absence of a sediment matrix. The presence of a sediment matrix corresponds to the scenario where fresh detritus particles are mixed in the deeper anoxic sediment as can occur physically through resuspension of sediment by turbulence in the bottom boundary layer (Ostrovsky et al. 1996, Ostrovsky and Yacobi 1999; Wüest and Lorke, 2003) and biologically through bioturbation by animals (Sun and Dai, 2005; Middelburg 2018). The absence of a sediment matrix corresponds to systems receiving a moderate to high organic matter load and where bottom water flow is not sufficient to induce resuspension (Kokic et al. 2016). In those systems, anoxia may develop and restrict bioturbation, thus neither physical nor biological mixing of the sediment will take place. The sediment and macrophytes were collected from tropical inland water because of the importance of these systems for global CH₄ emission (Tranvik et al. 2009; Bastviken et al. 2010).

Material and methods

Material collection

Macrophytes: The senescent aboveground tissues of nine different vascular aquatic plant species and one macroalgae (Table 1) were collected in four tropical lagoons with high macrophyte abundance and diversity (lagoons of Imboassica, Cabiúnas, Comprida and Carapebus, salinity <5.3 ppt, water depth <2.3 m, and total phosphorus (TP) concentration 0.36–1.28 µM; Caliman et al. 2010; Petruzzella et al. 2013) situated in the National Park of Jurubatiba in the state of Rio de Janeiro, Brazil. The entire aboveground tissues of several individuals (at least three) or only a part of them were used for the incubation, depending on the form of the macrophyte: aboveground tissues for *Ceratophyllum demersum* (Cera) and *Chara sp.* (Char), stems for *Eleocharis interstincta* (Elei) and *Eleocharis acutangula* (Elea), leaf for *Typha domingensis* (Typh), and leaf blade for the other species (see Table 1 for abbreviations). The senescent tissues collected were visibly beginning to decay due to their yellow/brown color and their quality was consequently assumed to be similar to that of the fresh detritus that is deposited on the sediment. The aboveground tissues were washed with tap water to remove sediment and invertebrates, cut to ca. 1 cm² and mixed.

Inoculum: One sediment core was sampled in each of the four lagoons of the macrophyte collection, and the top 10 cm of the four cores were mixed in equivalent proportions to constitute an inoculum. This inoculum was added to all slurries to ensure that a comparable microbial community containing methanogens was initially present in all treatments.

Sediment: Sediment was sampled in an oligotrophic drinking water reservoir (Chapeu d'Uvas) situated in the subtropical Atlantic Forest region of Brazil. The top 5 cm of three sediment cores sampled with a gravity corer (UWITEC, Austria) were kept after slicing, mixed, and stored in a closed bottle in the dark at 22°C, which is close to in situ temperatures. A previous

Table 1. Macrophyte sampled and characteristics (water content, C/N, and C/P) of the aboveground tissues used for the incubation.

Genus/species	Abbreviation	Family	Life form	Leaf water content (% of fresh weight)	C/N	C/P
<i>Chara sp.</i>	Char	Characeae	S	92 ± 0.5	11.2	376
<i>Ceratophyllum demersum</i>	Cera	Ceratophyllaceae	S	94.2 ± 0.6	16.2	—
<i>Nymphaea ampla</i>	Nyma	Menyanthaceae	FA	92.5 ± 0.3	23.1	968
<i>Nymphoides indica</i>	Nymi	Menyanthaceae	FA	92.9 ± 0	29.5	1436
<i>Potamogeton stenostachys</i>	Pota	Potamogetonaceae	FA	82.5 ± 1.4	30.2	2140
<i>Eichhornia crassipes</i>	Eicc	Pontederiaceae	FF	85.7 ± 0.9	43.1	1977
<i>Eichhornia azurea</i>	Eica	Pontederiaceae	FF/E	82 ± 1.1	49.8	2385
<i>Eleocharis interstincta</i>	Elei	Cyperaceae	E	91.6 ± 1.7	78	13,466
<i>Eleocharis acutangula</i>	Elea	Cyperaceae	E	91.8 ± 1.1	62.9	2593
<i>Typha domingensis</i>	Typh	Typhaceae	E	85.9 ± 3.4	89.9	3204

E, emergent plant; FA, floating leaved plant attached to the substrate; FF, free floating plant on water surface; S, submerged plant.

$n = 2$ for TOC and TN, and 3 for TP. C/N and C/P are molar ratios. The maximum standard deviations were 1% for TOC, 0.04% for TN, and 0.17 mg g⁻¹ for TP.

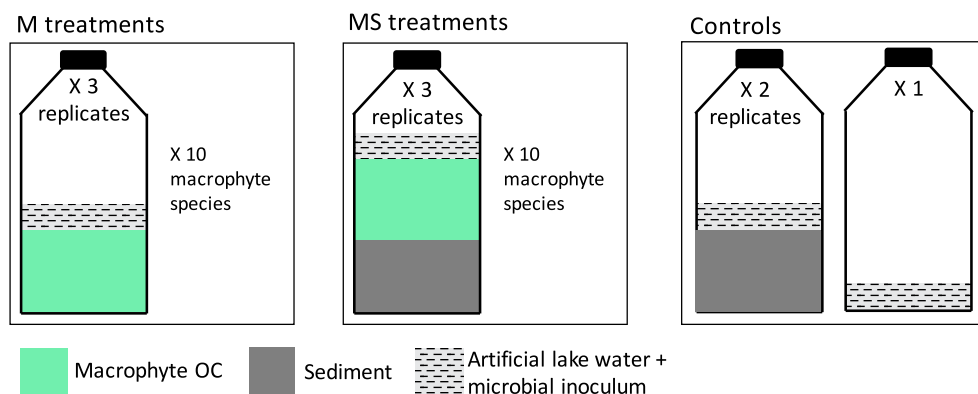
experiment with sediment collected in the same area showed that it has favorable conditions for methanogenesis (low redox potential, neutral pH) as well as a low CO₂ and CH₄ production at anoxic conditions (Grasset et al. 2018). We consequently used this nonsaline oligotrophic sediment for our incubation to ensure that few alternative electron acceptors would delay CH₄ production, and that most of the CH₄ would be derived from added organic matter.

Artificial lake water: Artificial lake water enriched in total nitrogen (TN, 4.57 mg L⁻¹ of NH₄NO₃) and TP (15.8 µg L⁻¹ of KH₂PO₄) was prepared according to Attermeyer et al. (2014), and used in all treatments to suspend the sediment and the macrophyte detritus.

All materials were stored in the dark at 4°C before the start of the incubation, and incubated fresh. The macrophytes and the inoculum were collected 2–3 d before the start of the incubation and the sediment was collected 1 month before the experiment.

Preparation of treatments: The incubation of each macrophyte species consisted of two treatments (M: macrophytes; MS: macrophytes and sediment) and was run as slurries. All

treatments contained macrophyte material from one of the 10 different species (1.0–2.6 g of fresh material corresponding to 44 to 80 mgC), a few drops of the inoculum (≈7 mgC) and 30 mL of artificial lake water. To the M treatments, no sediment was added, while the MS treatment included in addition 4.0–5.0 g of sediment (corresponding to 25–27 mgC). In the MS treatments, the high sediment to macrophyte OC ratio simulated an efficient surface sediment mixing. In the M treatments, only few sediment particles were added by the inoculum, and the low sediment to macrophyte OC ratio simulated the decomposition of macrophyte detritus without sediment mixing. Each of the 10 macrophyte species had three replicate slurries for both treatments (M and MS) resulting in a total of 60 different slurries. In addition, one control contained sediment, artificial lake water, and the inoculum in two replicates and another one contained only artificial lake water and inoculum (Fig. 1). All slurries and controls were incubated in 100 mL glass serum bottles (Merck KGaA, Darmstadt, Germany) closed with gas-tight 10-mm thick bromobutyl-rubber septa (Apodan, Denmark) and aluminum crimp seals.

**Fig. 1.** Experimental scheme. MS treatments correspond to macrophytes mixed with sediment while M treatments correspond to macrophytes without sediment.

Analyses of the materials used for incubation

The same sediment and macrophytes as those used for the incubation were dried at 60°C for 24–72 h for total organic carbon (TOC), TN, and TP analyses. Before analysis, samples were manually ground to a fine powder with a mortar and a pestle, except for some fibrous plant samples, which were cut into small pieces with a scissor before grinding. Water content was calculated as follows:

$$\text{water content (\% of fresh weight)} = \left(\frac{\text{fresh weight} - \text{dry weight}}{\text{fresh weight}} \right) \times 100 \quad (1)$$

For TOC analysis, 20–50 mg of macrophyte material and 200–400 mg of sediment were analyzed by high-temperature catalytic oxidation with a Shimadzu TOC equipped with a solid combustion system (TOC/L ASI-L, SSM 5000). Prior to TOC measurement, the sediment samples were acidified with 1 mL of 80% phosphoric acid to remove carbonates. For TN measurement, about 5 mg of macrophyte material was encapsulated in tin capsules and analyzed by high-temperature catalytic oxidation with a COSTECH system 4010 elemental analyzer. TP was measured after acid-persulfate digestion at 120°C in an autoclave for 2 h (Nelson 1987), and the dissolved phosphate was then analyzed according to the colorimetric assay of Murphy and Riley (1962).

Anaerobic incubation and gas measurements

The anoxic incubations were conducted for 59 d in the dark at a temperature between 22°C and 24°C. Slurries were only briefly shaken before gas measurements, as mixing can affect methanogenesis (Dannenberg et al. 1997). Anoxic conditions were obtained by flushing all slurries with N₂ at day 0 for 20 min after closing the bottles (Grasset et al. 2018). The slurries were then flushed every week with N₂ for 15 min to restore atmospheric pressure and avoid methanogenesis inhibition, which can be caused by high concentrations of CH₄ or other volatile compounds such as sulfides (Magnusson 1993; Guérin et al. 2008).

For CO₂ and CH₄ concentration measurements, 2 mL of the headspace was sampled three times per week with a plastic syringe equipped with a three-way valve and injected in an Ultra-Portable Gas Analyzer (Los Gatos Research Inc., Mountain View, CA) according to Grasset et al. (2018). Briefly, the gas analyzer was equipped with a gas-tight custom-made sample inlet and ambient outdoor air connected to a CO₂ absorber was used as a carrier gas. Injections led to peaks that were integrated with the R software (R version 3.3.2, R Core Team 2016) using a user-defined function. The area of the peaks was converted into molar units using a calibration curve and the ideal gas law.

pH was measured with a benchtop pH meter (Micronal, B474) at day 0, i.e., before macrophyte material addition in the artificial lake water (pH 6.9) and in the artificial lake water

mixed with sediment (pH 6.9). pH was also measured at the end of the incubation for all treatments and for the controls. pH values were relatively stable for the MS treatments (average final values between 6.7 and 7.8) but varied widely for the M treatments (average final values between 4.5 and 8.3; Table S1 in Supporting Information). Therefore, pH during the experiment was calculated by making a linear interpolation of pH from the beginning to the end of the incubation for each replicate. The concentration of dissolved inorganic carbon (DIC) was estimated from interpolated pH, measured CO₂ concentrations in the headspace, and equilibrium constants (Stumm and Morgan 1996). According to our estimation, dissolved carbonates (HCO₃⁻ and CO₃²⁻) constituted less than 24% of the total CO₂ (i.e., sum of headspace CO₂ and water-phase DIC) except for two species in the M treatments (58% and 37% of total CO₂ for Char and Cera, respectively). However, these concentrations of dissolved carbonates are uncertain since they were approximated from linearly interpolated pH. We therefore chose to report TCO₂ production as the sum of headspace and water-phase CO₂ production (excluding dissolved carbonates) as a conservative measure of CO₂ production during degradation.

Flushing the slurry headspace with N₂ removed 90% of CO₂ and 97% of CH₄ in the headspace and in the water phase as the samples were stirred while flushing. Cumulative TCO₂ and CH₄ production was calculated by adding the amounts removed by flushing to the concentration measured after flushing, and are used throughout the manuscript. CH₄ production rates were calculated between two flushing events as the slope of the linear change in CH₄ concentrations (three measurement points) vs. time. A previous experiment using sediment collected from the same spot and different plant OC types, including one of the macrophyte species used in this experiment, demonstrated that the CH₄ produced during the anoxic decomposition of fresh OC added to the sediment was fueled exclusively by the added plant OC (Grasset et al. 2018). For mass balance calculations, we consequently assumed that CO₂ and CH₄ only originated from the degradation of macrophyte OC. The production of CH₄-C and TCO₂-C (in gC) was divided by the initial amount of macrophyte OC, noted C_i (in gC), and expressed as percentage, as a measure of the transformation efficiency of macrophyte OC to CH₄ and CO₂:

$$\text{CH}_4\text{-C or TCO}_2\text{-C (in \% of } C_i) = \frac{\text{CH}_4 \text{ or TCO}_2 \text{ (in gC)}}{C_i \text{ (in gC)}} \times 100 \quad (2)$$

In addition, the C loss during incubation was calculated as the sum of CH₄-C and TCO₂-C in percent of C_i. Hence, the C loss is conservatively estimated as it excludes particulate as well as dissolved OC and carbonates. As part of the CH₄ produced can be consumed by anaerobic oxidation, it is important to note that CH₄ production refers to the result of the balance between methanogenesis and anaerobic CH₄ oxidation. As the focus of this study was on CH₄ production, TCO₂ values were

mainly used for C loss calculation and are only briefly mentioned in the result section.

Statistical analyses

To compare CH₄ production over time between the different macrophytes and the different treatments (M or MS), a nonlinear mixed-effects model was used. The accumulation of CH₄ concentration over time during the anaerobic incubation of fresh detritus in batch reactors, soils, or sediments typically follows a logistic curve, because after an eventual lag-time, CH₄ production is initially limited by the colonization of the detritus particles by anaerobic microorganisms, and followed by a substrate limitation at the end of the incubation (Kankaala et al. 2003; Vavilin et al. 2008; Ye et al. 2016). Hence, a simple logistic model was chosen to describe CH₄ production over time (Pinheiro and Bates 2000; Kankaala et al. 2003):

$$\text{CH}_4(t) = \frac{\text{Asym}}{1 + \exp[(xmid - t)/scal]} \quad (3)$$

This model predicts three parameters represented in Fig. 2: *Asym* is the horizontal asymptote and thus corresponds to the total CH₄ production, i.e., the extent of OC transformed into CH₄. If *Asym* is expressed as percent of initial macrophyte OC content (*C_i*, see Eq. 2), it corresponds to the modeled transformation efficiency of macrophyte OC to CH₄. *xmid* is the *t* value at which CH₄(*t*) equals *Asym*/2 and corresponds to the inflection point of the logistic curve where CH₄ production rate is maximum. *scal* represents the distance on the *x*-axis between *xmid* and the point where CH₄(*t*) equals *Asym*/(1 + *e*⁻¹). *scal* describes how quickly CH₄ production reaches the total CH₄ production, and is therefore related to the speed of CH₄ production (Fig. 2). The maximum CH₄ production

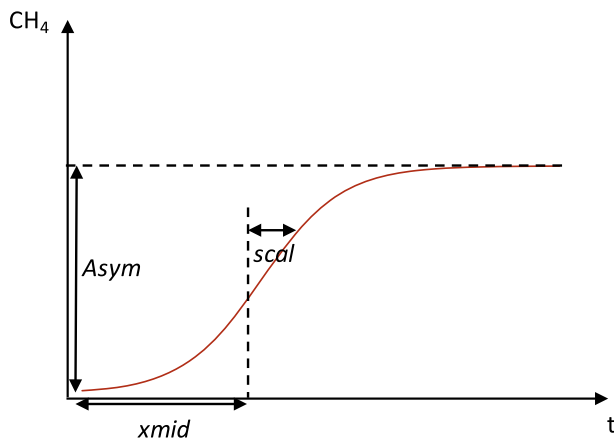


Fig. 2. The simple logistic model showing the parameters *Asym*, *xmid*, and *scal*, adapted from Pinheiro and Bates (2000) and describing CH₄ production. *Asym* corresponds to the total CH₄ production, i.e., the extent of OC transformed into CH₄, *scal* relates to the speed of CH₄ production, and *xmid* is the *t* value at which CH₄(*t*) equals *Asym*/2 at the inflection point where CH₄ production rate is maximum.

rate, noted *P*_{max}, can be estimated from *scal* and *Asym* according to the formula (Tsoularis and Wallace 2002):

$$P_{\max} = \left(\frac{d\text{CH}_4}{dt} \right)_{\max} = \frac{\text{Asym}}{4 \times \text{scal}} \quad (4)$$

As *P*_{max} integrates both the speed and the extent of CH₄ production, it can be considered as a measure of macrophyte OC reactivity.

The lag period was set to the period for which the amount of CH₄ produced was <2 μmol and was removed from the dataset for CH₄ modeling. CH₄ production was modeled using the self-starting function SSlogis, which calculates the starting parameters automatically, according to Pinheiro and Bates (2000). First, CH₄ production was modeled separately for the M and MS treatments to test if the model parameters (*Asym*, *xmid*, and *scal*) significantly differed between the different macrophytes. Time and the different macrophyte species were defined as fixed effects on the model parameters and the replicates per macrophyte were defined as random effects. Second, CH₄ production was modeled for M and MS treatments pooled together and the sediment presence was added as a fixed effect on the model parameters to compare the model parameters between M and MS treatments. Two macrophytes did not produce any CH₄ in the M treatments and could not be included in this second model, and this second model consequently included all data (M and MS treatments pooled) for the eight other macrophytes. The significance of the fixed and random effects on the model parameters was tested with the ANOVA function according to Pinheiro and Bates (2000). For the M treatments, the random effects for the parameter *scal* did not significantly improve the model and was therefore removed. The quality of the models was assessed by checking residuals and by plotting measured values against modeled values with the function “augPred” (Fig. S1 in Supporting Information; Pinheiro and Bates 2000).

Overall CH₄ production over time was well modeled by the simple logistic model, and the modeled transformation efficiency of macrophyte OC to CH₄ (i.e., parameter *Asym*) was in general close to the total CH₄ production measured at the end of the experiment (Fig. S1 and Table S2 in Supporting Information). However, for some treatments, the model slightly underestimated measured CH₄ production (Fig. S1 and Table S2 in Supporting Information). The estimated maximum CH₄ production rate (*P*_{max}) was very close to the highest production rate measured, showing again a good fit between the measured and modeled values (Table S2 and Fig. S2 in Supporting Information). As *Asym* and *P*_{max} were very close to the measured values and less affected by random error in single measurements, we used these modeled values to statistically compare the production of CH₄ over time between the different macrophytes and the different treatments.

The difference in C loss at the end of the experiment between M and MS treatments and the different macrophytes was tested with

a two-way ANOVA on the dataset excluding the two macrophytes which did not produce any CH₄ in the M treatments. The quality of the model was checked a posteriori with the normality and homoscedasticity of the residuals. The relationships between the macrophyte traits (water content, C/N, and C/P) and the model parameters (*Asym*, *scal*, and P_{\max}) and C loss were assessed with Spearman's rank correlations. All statistical analyses were performed with the R software.

Results

Measured CH₄ and CO₂ production, and C loss

The amounts of CH₄ produced were very low for the control without sediment and with sediment (5.8 and 10.4–11.0 μmol, respectively). The total CH₄ production measured at the end of the incubation was significant for all macrophytes except two, *Nymphoides indica* (Nymi) and *Eichhornia azurea* (Eica) in the M treatment. For those two macrophytes, the total CH₄ production measured was close to the limit of detection (0.8–1.5 μmol corresponding to a total CH₄ production of 0.01–0.03% of C_i ; Fig. 3, Table S2 in Supporting Information). The amounts of TCO₂ produced were also low for the control without sediment and with sediment (7.8 and 16.9–17.2 μmol, respectively). The amounts of TCO₂ produced were significant for all macrophytes in M and MS treatments (82.1–1017.8 μmol corresponding to a total TCO₂ production of 1.3–19.8% of C_i , Fig. S3 in Supporting Information). The measured total C loss (through CH₄ and TCO₂ production) at the end of the experiment varied between 1.28% ± 0.03% (Eica in M treatment) and 32.7% ± 4.1% (Nymi in the MS treatment), and was higher in the MS than in the M treatments ($p \leq 0.0001$, Fig. S4 in Supporting Information).

Differences in the modeled CH₄ production between macrophytes and correlation with plant traits

In both M and MS treatments, the modeled transformation efficiency to CH₄ (*Asym*) and speed of CH₄ production (*scal*) were highest for the submerged macrophytes (Char and Cera; Table 2). In the M treatments, *Asym* was 6.4% ± 0.6% and 8.8% ± 0.6% of C_i for Char and Cera, respectively, and on average 4.0% ± 2.9% of C_i for all macrophytes, including the two for which CH₄ production was equivalent to 0 (Table 2). In the MS treatments, *Asym* was 14.7% ± 1.1% and 14.6% ± 1.1% of C_i for Char and Cera, respectively, and on average 11.8% ± 2.9% of C_i for all macrophytes (Table 2). The estimated maximum CH₄ production rate (P_{\max}) was consistently high (>0.4% of C_i d⁻¹) for submerged species (Char and Cera), independently to the treatment. In the MS treatments, P_{\max} was also very high for Nymi (0.82% of C_i d⁻¹) and for *Nymphaea ampla* (Nyma) and Elea (ca. 0.4% of C_i d⁻¹) (Table 2), which are floating-leaved and emergent plants (Table 1).

The modeled transformation efficiency to CH₄, the speed of CH₄ production, and the estimated maximum CH₄ production rate (*Asym*, *scal*, and P_{\max} , respectively) correlated with the macrophyte water content, C/N, and C/P ratios (Table 3).

In particular, both *Asym* and P_{\max} were correlated negatively to C/N and positively to water content in the MS treatments (Fig. 4). In the MS treatments, macrophytes with a water content ≥92% (i.e., Char, Cera, Nyma, Nymi, Elei, and Elea) had a high P_{\max} (≥0.4% of C_i d⁻¹) except for Elei having high C/N and C/P values (Table 1). In the M treatments, macrophytes with a high water content also had a high or relatively high P_{\max} (between 0.24% and 0.51% of C_i d⁻¹) except for Elei and Nymi which produced no or very little CH₄ (Tables 1, 2).

Comparison of the modeled CH₄ production between M and MS treatments

For the two species Nymi and Eica, where no CH₄ production could be detected in the M treatment, we measured a significant CH₄ production in the MS treatment; in fact, Nymi had the highest maximum CH₄ production rate (P_{\max}) in the MS treatment (Table 2). For the other eight macrophytes, the model parameters *Asym* and *scal* were significantly different between M and MS treatments (p value of the fixed effect sediment <0.001 for both parameters; Table S3 in Supporting Information). The presence of sediment affected *Asym* and *scal* differently depending on the macrophyte (significant interaction sediment*macrophyte, $p \leq 0.01$ for *Asym* and *scal*; Table S3 in Supporting Information). The modeled transformation efficiency to CH₄ (*Asym*) in presence of sediment was a factor of 2–8 higher than in absence of sediment, while the speed of CH₄ production (*scal*) was mostly lower in the presence of sediment (Table 2). The lag time was also affected by the presence of sediment; it was longer in the M treatments (≥15 d for seven macrophytes) than in the MS treatments (2 d for eight macrophytes; Table 2).

Discussion

Differences in CH₄ production between macrophytes

The efficiency of plant OC transformation to CH₄ strongly differed among macrophyte species at anoxic conditions. The transformation efficiency to CH₄ varied between 0% and 15.0% of C_i (*Asym* in Table 2), and the interspecies differences were related to the macrophyte's water content and nutrient stoichiometry, thereby corroborating our initial hypothesis. Macrophytes with higher water content and lower C/N ratio produced more CH₄ during anoxic decomposition (correlation with *Asym*) and had higher estimated maximum CH₄ production rates (correlation with P_{\max} ; Table 3, Fig. 4). Several studies found that the transformation efficiency of OC into CH₄ could increase by 2–3-fold for some macrophyte species in comparison to others (Kankaala et al. 2003; Vizza et al. 2017) and was higher for algae than for terrestrial leaves (West et al. 2012). CH₄ production has been related to peat C/N content (Valentine et al. 1994) and phytoplankton lipid content (West et al. 2015) but no correlation with macrophyte species stoichiometry or water content has been found (Vizza et al. 2017). This study is consequently the first reporting systematic interspecies difference

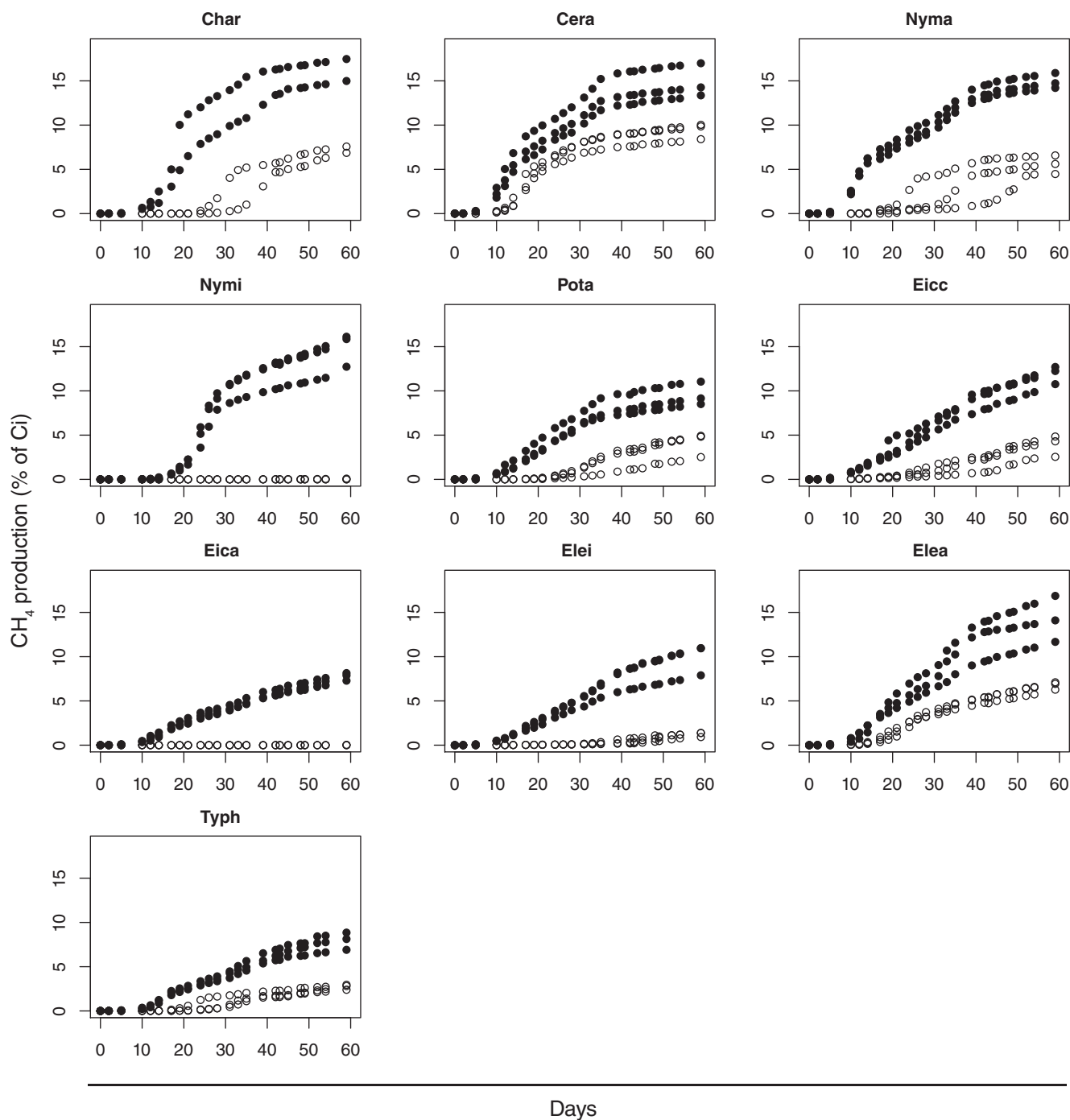


Fig. 3. CH_4 production over time, expressed as percent $\text{CH}_4\text{-C}$ of initial macrophyte OC, for the macrophyte detritus mixed with sediment, MS (points) and for the macrophyte alone, M (circles) treatments.

in macrophyte OC transformation efficiency to CH_4 coupled to C/N ratio and water content. C/N is often used as an indicator of organic matter lability, and a high C/N ratio indicates that organic matter is rich in complex compounds such as polysaccharides or lignins and that N might be limiting for microbial degradation (Duarte 1992; Enriquez et al. 1993). The correlation that we found between the C/N ratio

of macrophyte detritus and CH_4 production may consequently be attributed to a slow hydrolysis of complex compounds (Kristensen et al. 1995) or a low N content that can limit methanogenesis (Ferry 2012). In the same way, the leaf water content likely related to CH_4 production because it is inversely proportional to the abundance of structural compounds, compounds that can limit methanogenesis due

Table 2. Results of the simple logistic model of CH₄ production.

M treatments						
	Lag time (d)	Asym (% of C _i)	Level	scal (d)	Level	P _{max} (% of C _i d ⁻¹)
Char	23 ± 3	6.4 ± 0.6	A	3.1 ± 0.9	B	0.51
Cera	5 ± 0	8.8 ± 0.6	A	4.4 ± 0.9	B	0.5
Nyma	15 ± 3	5.6 ± 0.6	B	4.4 ± 0.9	B	0.32
Nymi	52 ± 0	0*	—	—	—	—
Pota	16 ± 2	4.1 ± 0.6	B	6.3 ± 1.0	B	0.17
Eicc	7 ± 3	5.0 ± 0.45	B	9.3 ± 0.9	A	0.13
Eica	54 ± 0	0*	—	—	—	—
Elei	16 ± 2	1.3 ± 0.7	C	5.8 ± 2.2	A	0.06
Elea	7 ± 2	6.1 ± 0.6	B	6.5 ± 0.9	B	0.24
Typh	16 ± 3	2.5 ± 0.6	C	5.8 ± 1.2	B	0.11

MS treatments						
	Lag time (d)	Asym (% of C _i)	Level	scal (d)	Level	P _{max} (% of C _i d ⁻¹)
Char	4 ± 1	14.7 ± 1.1	A	5.8 ± 0.6	B	0.64
Cera	2 ± 0	14.6 ± 1.1	A	7.6 ± 0.7	B	0.48
Nyma	2 ± 0	15.0 ± 1.1	A	9.4 ± 0.7	A	0.4
Nymi	11 ± 1	13.0 ± 1.1	B	4.0 ± 0.6	B	0.82
Pota	2 ± 0	9.2 ± 1.1	C	6.6 ± 0.7	B	0.34
Eicc	2 ± 0	11.7 ± 0.7	B	9.2 ± 0.5	A	0.32
Eica	2 ± 0	7.6 ± 1.1	C	9.8 ± 1.0	A	0.19
Elei	2 ± 0	9.9 ± 1.1	B	8.7 ± 0.8	A	0.29
Elea	2 ± 0	14.4 ± 1.2	A	8.8 ± 0.8	A	0.41
Typh	2 ± 0	8.2 ± 1.1	C	9.5 ± 1.0	A	0.22

The model parameter *Asym* corresponds to the transformation efficiency of macrophyte OC to CH₄, and *scal* relates to the speed of CH₄ production: the lower the *scal* is, the quicker the total CH₄ production is reached. The estimated maximum CH₄ production rate (*P*_{max}) is calculated as $P_{max} = \frac{Asym}{4 \times scal}$, thus it integrates both the speed and the extent of CH₄ production and relates to macrophyte OC reactivity.

The different levels are given with the species Eicc as the reference level, which was chosen because it is of intermediate reactivity, enabling to distinguish very reactive macrophyte OC from relatively unreactive macrophyte OC. A different letter represents a significantly higher (A) or lower (C) value of the model parameter than that of Eicc.

The lag time is given in mean ± SD and the model parameters *Asym* and *scal* are given in mean ± SE.

*Two macrophyte did not produce CH₄ in the M treatments and could not be included in the model, *Asym* was considered equivalent to 0 for calculating averages.

to a slow hydrolysis. Therefore, the macrophyte's water content and C/N ratio can provide predictive ranges of the transformation efficiency to CH₄ and the maximum CH₄ production rate (Table 4). Future studies should refine the relationships between water content, C/N ratio, and CH₄ production in order to model more accurately CH₄ production over time for different macrophyte species.

The two submerged macrophytes had among the highest modeled transformation efficiency to CH₄ and speed of CH₄ production (Table 2). However, the speed and the extent of OC transformation to CH₄ varied widely between floating species and we did not find significant differences between the other life forms (i.e., floating and emergent species, Tables 1, 2). Descriptors related to macrophyte lability and in particular water content and C/N ratio seemed consequently more accurate than the different life forms to describe the CH₄ production potential of macrophyte detritus.

Since most macrophyte OC is transformed to CH₄ at a short time scale (<100 d, Kankaala et al. 2003; Grasset et al. 2018), the transformation efficiency to CH₄ given by our short-term

Table 3. Spearman coefficients of the correlations between modeled parameters of CH₄ production (*scal*, *Asym*, and *P*_{max}), C loss, and the plant traits (C/N, water content, and C/P). The significant correlations among *Asym*, *P*_{max}, and the plant traits are represented in Fig. 4 for the MS treatments.

	M			MS		
	C/N	Water	C/P	C/N	Water	C/P
<i>Asym</i>	-0.79*	ns	-0.79*	-0.72*	0.81**	ns
<i>scal</i>	ns	-0.72*	ns	ns	ns	ns
C loss	ns	ns	ns	ns	0.84**	ns
<i>P</i> _{max}	-0.90**	ns	-0.86*	-0.73*	0.81**	ns

ns, not significant. ** *p* < 0.01; * *p* < 0.05.

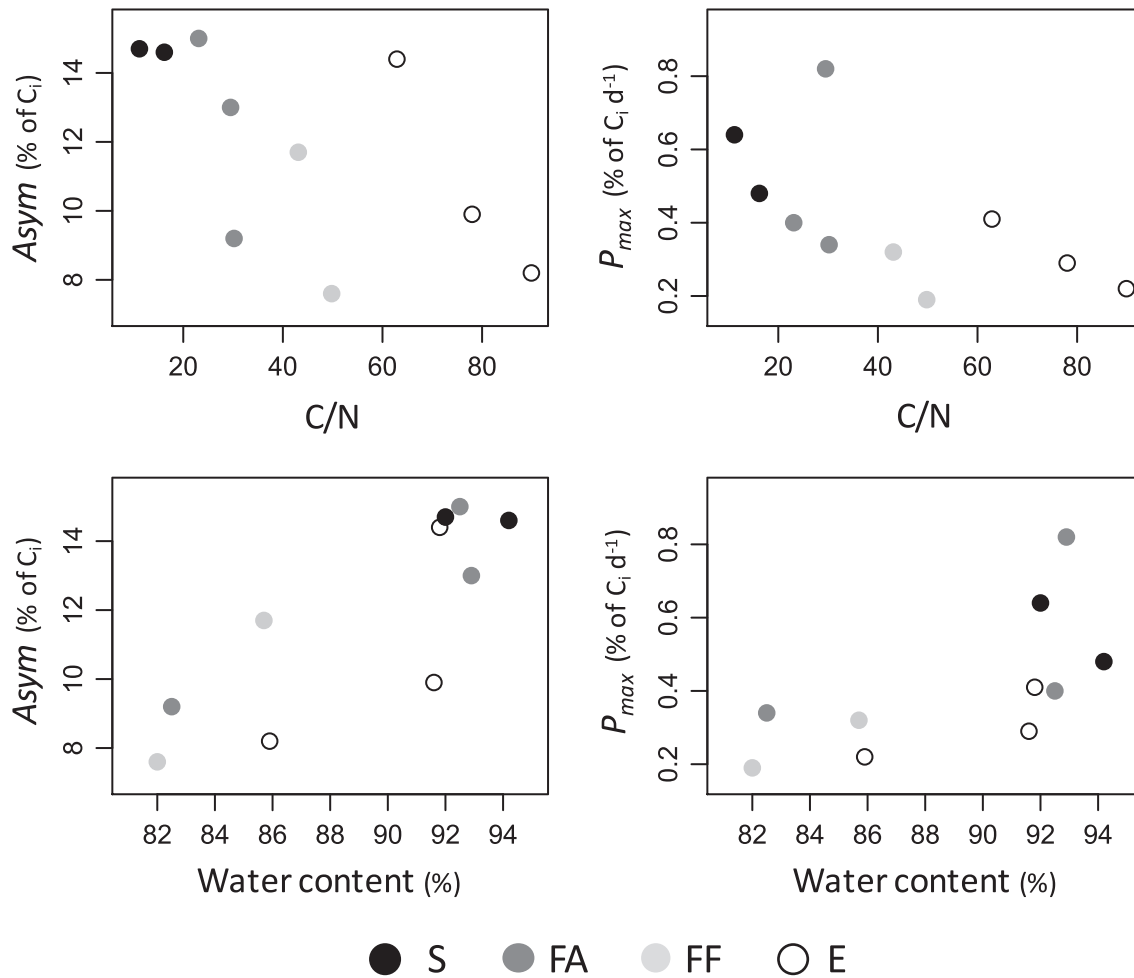


Fig. 4. Significant correlations among $Asym$, P_{max} , and plant traits (C/N, water content) during the degradation of macrophytes mixed with sediments (MS treatments). S submerged (black circles), FA floating attached to the substrate (gray), FF free floating (light gray), and E emergent (white). Eica is represented here as free floating but it can also have the other life form emergent (Table 1). See Table 3 for Spearman correlation coefficients and p -value levels.

experiment represents the majority of CH_4 produced. However, a part of the OC will continue to decompose at slow rates over longer time scales (years or decades) and fuel CH_4 production in deeper sediment layer (Gebert et al. 2006; Sobek et al. 2012). Furthermore, other factors than the quality of OC, such as pH, microbial communities, or competitive electron acceptor content (Valentine et al. 1994) are known to affect CH_4 production. It would be consequently interesting to study the decomposition of macrophytes in different anoxic sediments to test how these factors can affect the transformation efficiency to CH_4 and the maximum CH_4 production rate.

Effect of sediment presence on CH_4 production from macrophyte detritus

Our results show that the presence of sediment strongly affected CH_4 production from macrophyte detritus: the transformation efficiency of OC to CH_4 ($Asym$) was higher if the macrophyte detritus was mixed with sediment (MS treatments) than not (M treatments) (Fig. 3, Table 2). The values of total

CH_4 production were consistent with the literature, Kankaala et al. (2003) found a total CH_4 production of 5–17% of C_i for macrophyte detritus decomposing without sediment ($Asym$ between 0% and 8.8% of C_i for M treatments in the present study), and

Table 4. Predictive ranges of the transformation efficiency of macrophyte OC to CH_4 ($Asym$, % of C_i) and maximum production rate (P_{max} , % of C_i d⁻¹) during the anoxic degradation of macrophyte detritus, according to the macrophyte water content and C/N ratio.

	Not mixed with sediment		Mixed with sediment	
	$Asym$	P_{max}	$Asym$	P_{max}
Water content $\geq 92\%$ and C/N < 63	6–9*	0.2–0.5*	13–15	0.4–0.8
Water content $< 92\%$ or C/N > 63	1–5*	0.1–0.2*	8–12	0.2–0.3

*The two macrophytes (Nymi and Eica) that did not produce CH_4 when not mixed with sediment are excluded from these ranges.

Grasset et al. (2018) found a CH₄ production of 7–20% of C_i for macrophytes mixed with sediments (*Asym* between 7.6% and 15.0% of C_i for MS treatments in the present study). The consistently higher CH₄ production for the MS treatments compared to the M treatments supports our initial hypothesis, and we attribute this difference to physicochemical conditions favorable for CH₄ production (low redox potential buffered pH). Furthermore, sediment mineral surfaces can enhance biofilm formation, favor interactions between methanogenic consortia, and thereby ultimately stimulate CH₄ production (Sanchez et al. 1994; Tolker-Nielsen and Molin 2000).

The difference in total CH₄ production between M and MS treatments varied between the macrophyte species in relation to pH. The absence of CH₄ production for *Eica* and *Nymi* in the M treatments was concomitant with a low final pH (pH of 4.5 ± 0.2 and 5.0 ± 0.2 for *Nymi* and *Eica* in the M treatment, respectively). Similarly, there was little CH₄ production in another relatively acidic treatment (pH of 5.8 ± 0.7 for *Elei* in the M treatment; Table S1 in Supporting Information), but higher CH₄ production for all other macrophytes in the M treatments where pH was ≥7. While this suggests that low pH due to plant decay may cause an inhibition of methanogenesis, several studies found contradictory results on the importance of pH for CH₄ production (Deano and Robinson 1985; Valentine et al. 1994). For example, it is also possible that the low pH is the result of methanogenesis inhibition as an unbalanced acidogenesis can lead to low pH due to fatty acids accumulation (Franke-Whittle et al. 2014). Several compounds contained in macrophyte tissues or formed during their decomposition could cause methanogenesis inhibition such as phenols or fatty acids (Chen et al. 2008; Emilson et al. 2018), and it is consequently not possible to conclude on the cause of methanogenesis inhibition. Our study suggests that in the case of a very high load of macrophyte detritus deposited on top of the sediment at anoxic conditions, as could happen in productive sites with calm waters (e.g., wind-protected littoral zones of lakes and wetlands), some macrophyte species might not decompose to any large extent, and produce comparatively little CH₄. When judging the extent of CH₄ production from macrophytes, it is consequently important to consider how much the macrophyte OC is mixed with sediment.

While increasing the total CH₄ production, the presence of sediment reduced the speed of CH₄ production (*scal* in Table 2). The slower OC decomposition in presence of sediment may be attributed to a slower diffusion rate of enzymes within the sediment matrix because the high tortuosity of sediments increases diffusion distances and lowers the accessibility of the OC to enzymatic attack (Rothman and Forney 2007). The higher C loss combined with the slower OC decomposition rate in the MS treatments may also indicate that organic compounds of lower degradability and thus with potentially slow hydrolysis or fermentation rates (Kristensen et al. 1995; Bastviken et al. 2003) could be degraded in presence of sediment (Fig. S4 in Supporting Information). Furthermore, it is possible that other anaerobic

pathways of potentially different OC mineralization rates, such as iron reduction, might be involved in presence of sediment (Lovley 1987; Quintana et al. 2015).

Implications

According to our study, macrophytes with low C/N ratio and high water content have the potential to induce high CH₄ emissions, in cases where the macrophyte detritus decomposes anoxically and a significant fraction of the produced CH₄ escapes oxidation and is delivered to the atmosphere. Both the speed and the extent of OC transformation to CH₄ are important with respect to eventual emission of CH₄ from a sediment. A high CH₄ production rate is more likely to lead to CH₄ bubble formation and effective transport of CH₄ via bubbles from sediment to the atmosphere (ebullition), because CH₄ oversaturation in sediment pore water is reached rapidly if the rate of CH₄ production greatly exceeds the rate of CH₄ diffusion from the sediment to the water column. Conversely, with a slow rate of CH₄ production, CH₄ oversaturation is unlikely to be reached, CH₄ will leave the sediment slowly via diffusion, and a large proportion of the CH₄ diffusing from sediments will be oxidized to CO₂ (Chanton and Whiting 1995; Bastviken 2009; Sobek et al. 2012). These findings suggest that macrophytes with high water content and low C/N ratio, such as the two submerged macrophytes *Chara* and *Ceratophyllum*, have the potential to trigger high CH₄ production rates and CH₄ bubble formation in the sediment, and ultimately CH₄ emission through ebullition. It is however important to consider that CH₄ production rates and the release of bubbles depend on several other environmental factors (e.g., temperature, hydrostatic, or atmospheric pressure, Mattson and Likens 1990; Yvon-Durocher et al. 2014), and of course the oxygenation regime. The quantity of CH₄ delivered to the atmosphere will also depend on the fraction that is transported via the plant aerenchyma as this pathway bypasses CH₄ oxidation (Schütz et al. 1991; Chanton and Whiting 1995). Some rooted floating macrophytes (e.g., *Nymphaea sp.* and *Nymphoides sp.*) that have a high CH₄ production potential according to our study also have the capacity to efficiently transport CH₄ to the atmosphere through their tissues (Grosse and Mevi-Schutz 1987; Schütz et al. 1991). The anoxic decomposition of these macrophytes could consequently result in high CH₄ emissions. On the other hand, for rooted macrophytes, the fraction of CH₄ that is lost by oxidation in the plant root vicinity can also be important (Laanbroek, 2010; Ribaud et al. 2012). To have a comprehensive understanding on the effect of different macrophyte species on CH₄ emissions and to model CH₄ emissions at an ecosystem scale, it would be necessary to quantify how much CH₄ produced by macrophyte detritus is transported through the plant, emitted via ebullition or oxidized by methanotrophs living in the rhizosphere, given that these processes can differ between plant species (Ström et al. 2003; Bhullar et al. 2013; Yoshida et al. 2014). Our study is a first step toward modeling CH₄ emissions at an ecosystem scale since it relates CH₄ production to macrophyte traits (C/N ratio and

water content) and shows that the environment in which the macrophyte detritus is deposited (mixed into the sediment, or deposited on top of the sediment) affects the rate and extent of CH₄ production.

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Acknowledgments

The authors thank Leticia Weber, Gabriella Villamor Saucedo, and Lucas Bolenger for help with laboratory work, Birgit Koehler for advices regarding statistical analyses, and Kevin Pailhes for help during fieldwork. The authors also thank NUPEM/UFRJ and ICMBIO (National Park of Jurubatiba) for logistic support on site. The research leading to these results has received funding from the European Research Council under the European Union's Seventh Framework Programme (FP7/2007-2013)/ERC grant agreement no. 336642 and from CNPq (National Council for Scientific and Technological Development)/PVE project n8 401384/2014-4. FR was partially supported by CNPq.

Conflict of Interest

None declared.

Submitted 27 July 2018

Revised 21 December 2018

Accepted 29 January 2019

Associate editor: Kimberly Wickland